

Name: \_\_\_\_\_

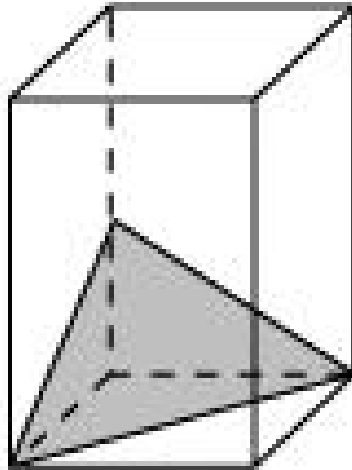
Geos 306  
2<sup>nd</sup> 2008 Midterm (100 pts)

1) (12 pts) Fill in the table that defines the shape and constraints of the unit cell parameters.

System	a	b	c	$\alpha$	$\beta$	$\gamma$
Cubic						
Hexagonal						
Tetragonal						
Orthorhombic						
Monoclinic						
Triclinic						

- 2) (3 pts) Draw a picture of a tetragonal lattice projected onto the  $a$ - $b$  plane (i.e. you do not need to show the  $c$ -direction); indicate the origin and the cell edges. Then draw the vector  $[2 \ -1 \ 0]$ .

- 3) (5 pts) In the picture below is an image of a unit cell and a plane. Designate an origin, and label 3 edges as  $\mathbf{a}$ ,  $\mathbf{b}$ , and  $\mathbf{c}$ . What are the Miller indices of the indicated plane?



- 4) (30 pts) Fill in the missing point group symbols and indicate the crystal system for the point groups on a given row.

G	GUGi	HUG\Hi	System
1		n/a	
	2/m		
3		n/a	
	4/m		
		$\bar{6}$	
		2mm	
	$\bar{3}2/m2/m$		
		4mm and $\bar{4}2m$	
		6mm and $\bar{6}2m$	
332		n/a	
	4/m $\bar{3}$ 2/m		

5) (10 pts) Make a careful drawing and derive Bragg's law for X-ray diffraction.

- 6) (10 pts) Consider a crystal with tetragonal symmetry,  $a = 4.177 \text{ \AA}$ ,  $c = 2.665 \text{ \AA}$ .
- (a) Draw a figure showing the plane (110).
  - (b) Compute the d-spacing of (110)?
  - (c) Using Cu radiation,  $\lambda = 1.54 \text{ \AA}$ , what position in  $2\theta$  would the (110) peak be found.

- 7) (15 pts) The density of SiO<sub>2</sub> quartz is 2.64 g/cm<sup>3</sup>. Cristobalite is a high-temperature phase of silica, SiO<sub>2</sub>, found in volcanic rocks. It is tetragonal with cell edges  $a = 4.9717 \text{ \AA}$ ,  $c = 6.9223 \text{ \AA}$  and  $Z = 4$  (i.e. 4 formula units per unit cell). Recall that Avogadro's number is  $6.023 \times 10^{23} \text{ mole}^{-1}$  and that  $10^8 \text{ \AA} = 1 \text{ cm}$ .
- Determine the cell volume of cristobalite.
  - Determine the molecular weight of SiO<sub>2</sub>.
  - Determine the density of cristobalite in units of gm/cm<sup>3</sup>.
  - Why should the density of cristobalite be greater or less than quartz?

8) (15 pts) From the microprobe analyses of a mineral, the following oxide weight percents were measured. Determine the chemical composition of this sample.

	Weight percent			
MgO	30.00			
Al <sub>2</sub> O <sub>3</sub>	25.29			
SiO <sub>2</sub>	44.71			

Bonus point: name the mineral!